

Fig. 5. GC/electron-capture detector of the pooled plasma (6 h postdosage); see Fig. 4A for the same sample on MPD and note the simplified chromatogram using the MPD-specific fluorine detector. (A) Rat plasma, 6 h postdosage; (B) A spiked with the unsaturate (C<sub>2</sub>F<sub>15</sub>CF=CHCOO<sup>-</sup>). (w) 2-Hydroperfluoro-2decenate (C<sub>7</sub>F<sub>15</sub>CF=CHCOO<sup>-</sup>); (x) perfluorooctanoate (C<sub>7</sub>F<sub>15</sub>COO<sup>-</sup>); (y) 2H,2H-perfluorodecanoate (C<sub>8</sub>F<sub>17</sub>CH<sub>2</sub>COO<sup>-</sup>). Note the unidentified fluorine containing metabolite (z) is not apparent with the electroncapture detector. Furthermore, while the MPD responds to the total fluorine content of a compound per the empirical formula, the electron-capture detector has different sensitivities for different compounds. In this series, the sensitivity is C<sub>7</sub>F<sub>15</sub>CF=CHCOOCH,  $> C_1F_1 \cdot CF_2CH_2COOCH_1 > C_2F_1 \cdot CF_2CH_2CH_2OH.$ 

The fluorine-containing biotransformation product of the alcohol eluting at 10.0 min has not as yet been identified but it is speculated that it contains a carboxyl group since diazomethane renders it volatile for GC. The unsaturate has been identified (by GC retention data) and an additional metals.

abolic derivative is possible in that the small est molecular metabolite (C<sub>7</sub>F<sub>15</sub>COO<sup>7</sup>) observed in this work has one less CF<sub>2</sub> group than the starting alcohol.

The metabolism of fluorine-containing and perfluoro long-chain acids has not been studied in great detail. It is known that the perfluoroheptyl chain (specifically perfluorocanoate) is metabolically stable (14-16) while ω-fluorocarboxylic acids (17) undergo β oxidation. The high inorganic fluoride level in the plasma (Table 1) and formation of perfluorocatanoate suggest the overall re-

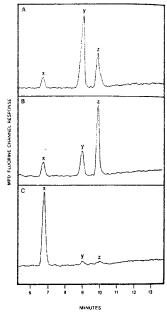


Fig. 6. Rat plasma extracts at various times, post-dosage. (A) Rat 10 (2 h postdosage; (B) Rat 9 (6 h postdosage); (C) Rat 20 (48 h postdosage). (x) Perfluorococtanoate (C<sub>2</sub>F<sub>1</sub>,COO<sup>-</sup>); (y) 2H<sub>2</sub>H<sub>2</sub>perfluore-decanoate (C<sub>3</sub>F<sub>1</sub>,CH<sub>2</sub>COO<sup>-</sup>); (z) unidentified metabolite

## CF<sub>18</sub>CF<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OH — #1#.2#.2#-perfluorodecanol

 $C_7F_{15}CO_2^- + 2 HF$ 

Figure 5 illustrates the electron capture issponse obtained for the esterified plasma extract on the capillary column system and shows the complex mixture which is simplified by monitoring only the fluorine content on the GC/MPD. It is obvious that an ideal system would combine the selectivity of the microwave-sustained helium plasma detector with the improved separation capabilities of a capillary system and this is under development at this time.

MPD plots of the fluorine channel response are shown for various rats (Fig. 6). Note that the 2H,2H-perfluorodecanoic acid ester is the predominant component in the 2-h-postdosage rat. The chromatograms in Fig. 6 illustrate the progressive biotransformation of the alcohol to perfluorooctanoate. Differences in the rate of biotransformation of the alcohol between rats were observed.

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